



# RJ VISION PVT. LTD.

Code  
**A**

## RJ – CET (2015 – 16)

(Entrance cum Scholarship Test)

### COURSE – RADIANT (CLASS 12<sup>TH</sup>)

(Test Syllabus - Class 11<sup>th</sup>)

TIME : 1.15 HR MM : 180

This Booklet contains \_\_\_\_\_ pages

#### Important Instructions :

1. The test is of **1 hour** duration and Test Booklet contains **60** questions. Each question carries **4** marks. For each correct response, the candidate will get **4** marks. For each incorrect response, **one mark** will be deducted from the total scores. The maximum marks are **240**.
2. Use **Blue/Black Ball point pen only** for writing particulars on this page/markings responses.
3. Rough work is to be done on the space provided for this purpose in the Test Booklet only.
4. **On completion of the test, the candidate must handover the Answer Sheet to the invigilator in the Room/Hall.**
5. Use of white fluid for correction is **NOT** permissible on the Answer Sheet.
6. Each candidate, must show on demand his/her Admission Card to the Invigilator.
7. No candidate, without special permission of the Superintendent or Invigilator, would leave his/her seat.
8. Use of Electronic/Manual Calculator is prohibited.
9. No part of the Booklet and Answer Sheet shall be detached under any circumstances.
10. The candidates will write the Correct Test Booklet Code as given in the Test Booklet/Answer Sheet Attendance Sheet.

Do not open this Test Booklet until you are asked to do so.

Name of the Candidate (in Capitals) : \_\_\_\_\_

Roll Number : \_\_\_\_\_

School : \_\_\_\_\_

Centre of Examination (in Capitals) (Vasna / Niz / Anand / Others) \_\_\_\_\_

Candidate's Signature : \_\_\_\_\_ Invigilator Signature : \_\_\_\_\_

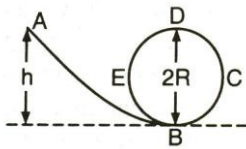
1. The relation between time  $t$  and distance  $x$  is:

$$t = \alpha x^2 + \beta x$$

Where  $\alpha$  and  $\beta$  are constants. The retardation is:

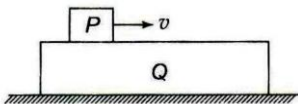
- (1)  $2\alpha v^3$                       (2)  $2\beta v^3$   
 (3)  $2\alpha\beta v^3$                     (4)  $2\beta^2 v^3$

2. A frictionless track ABCDE ends in a circular loop of radius  $R$ . A body slides down the track from point A which is at a height  $h = 5$  cm. Maximum value of  $R$  for the body to successfully complete the loop is :



- (1) 5 cm    (2)  $\frac{15}{4}$  cm    (3)  $\frac{10}{3}$  cm    (4) 2 cm

3. A block P of mass  $m$  is placed over a plank of mass  $M$ . Plank Q is placed over a smooth horizontal surface as shown in Fig. Block P is given a velocity  $v$  to the right. If  $\mu$  is the coefficient of friction between P and Q, the acceleration of Q relative to P is



- (1)  $\mu g$                       (2)  $\frac{\mu mg}{M}$   
 (3)  $\mu g \left(1 - \frac{m}{M}\right)$     (4)  $\mu g \left(1 + \frac{m}{M}\right)$

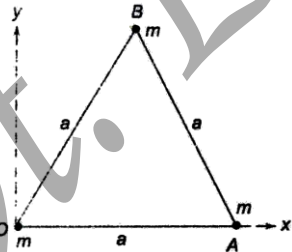
4. A body of mass  $m$ , having momentum  $p$ , is moving on a rough horizontal surface. If it is stopped in a distance  $x$ , the coefficient of friction between the body and the surface is given by

- (1)  $\mu = \frac{p^2}{2gm^2x}$                       (2)  $\mu = \frac{p^2}{2mgx}$   
 (3)  $\mu = \frac{p}{2mgx}$                       (4)  $\mu = \frac{p}{2gm^2x}$

5. A body of mass  $m$  moving with a certain speed suffers an inelastic collision with a body of mass  $M$  at rest. The ratio of the final kinetic energy of the system to the initial kinetic energy is

- (1)  $\frac{m}{m+M}$                       (2)  $\frac{M}{m+M}$   
 (3)  $\frac{m+M}{m}$                       (4)  $\frac{m+M}{M}$

6. Three particles each of mass  $m$ , are placed at the corners of an equilateral triangle of side  $a$ , as shown in fig. The position vector of the centre of mass is



- (1)  $\frac{a}{2}(i + j/\sqrt{3})$                       (2)  $\frac{a}{2}(3i + j)$   
 (3)  $\frac{a}{2}(3i + \sqrt{3}j)$                       (4)  $\frac{a}{2}(3i + j/\sqrt{3})$

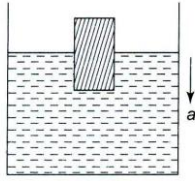
7. If  $g$  is the acceleration due to gravity on the surface of the earth, the gain in potential energy of a satellite of mass  $m$  raised from the earth's surface to a height equal to the radius  $R$  of the earth is

- (1)  $mgR/4$   
 (2)  $mgR/2$   
 (3)  $mgR$   
 (4)  $2 mgR$

8. A wire of length  $L$  and cross-sectional area  $A$  is made of a material of Young's modulus  $Y$ . The work done in stretching the wire by an amount  $x$  is given by

- (1)  $\frac{YAx^2}{L}$                       (2)  $\frac{YAx^2}{2L}$   
 (3)  $\frac{YAL^2}{x}$                       (4)  $\frac{YAL^2}{2x}$

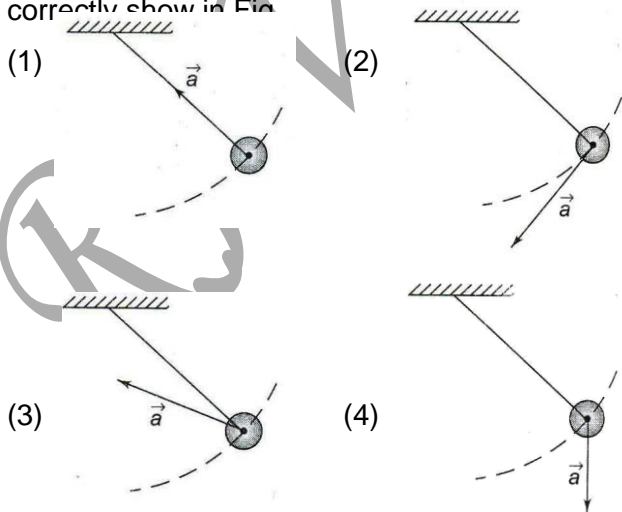
9. A block floats in a liquid contained in a beaker (Fig). The beaker is placed on the floor of an elevator. If the elevator descends with acceleration  $a$  ( $< g$ ), the upthrust on the block due to the liquid



- (1) is equal to the weight of the liquid displaced  
 (2) is greater than the weight of the liquid displaced  
 (3) is less than the weight of the liquid displaced  
 (4) become equal to zero
10. A uniform rod AB of length  $L$  is pivoted at one end A and hangs vertically. The time period of small oscillations about an axis passing through A and perpendicular to the rod is

- (1)  $2\pi\sqrt{\frac{L}{g}}$                       (2)  $2\pi\sqrt{\frac{L}{2g}}$   
 (3)  $2\pi\sqrt{\frac{L}{3g}}$                       (4)  $\pi\sqrt{\frac{L}{2g}}$

11. A simple pendulum is oscillating without damping. When the displacement of the bob is less than maximum, its acceleration vector is correctly show in Fig



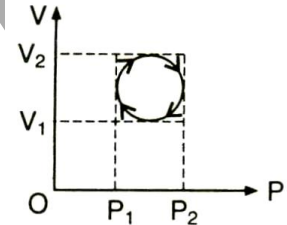
12. Two interfering waves have intensities in the ratio 9 : 1. Then the ratio of maximum to minimum amplitude is :

- (1) 10 : 8  
 (2) 4 : 2  
 (3) 100 : 64  
 (4) 16 : 4

13. If heat energy  $\Delta Q$  is supplied to an ideal diatomic gas, the increase in internal energy is  $\Delta U$  and the work done by the gas is  $\Delta W$ . The ratio  $\Delta Q : \Delta U : \Delta W$  is

- (1) 5 : 3 : 2                      (2) 5 : 2 : 3  
 (3) 7 : 5 : 2                      (4) 7 : 2 : 5

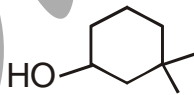
14. In the cyclic process shown on the P – V diagram the magnitude of the work done is :



- (1)  $\pi\left(\frac{P_2 - P_1}{2}\right)^2$   
 (2)  $\pi\left(\frac{V_2 - V_1}{2}\right)^2$   
 (3)  $\frac{\pi}{4}(P_2 - P_1)(V_2 - V_1)$   
 (4)  $\pi(P_2V_2 - P_1V_1)$

15. The root mean square velocity,  $v_{rms}$ , the average velocity  $v_{av}$  and the most probable velocity,  $v_{mp}$  of the molecules of the gas are in the order :

- (1)  $v_{mp} > v_{av} > v_{rms}$   
 (2)  $v_{rms} > v_{av} > v_{mp}$   
 (3)  $v_{av} > v_{mp} > v_{rms}$   
 (4)  $v_{mp} > v_{rms} > v_{av}$

16. Number of nodal centres for 2s orbital  
(1) 1 (2) 0 (3) 4 (4) 3
17. The orbital angular momentum of an electron in 2s-orbital is  
(1)  $\frac{1}{2} \frac{h}{2\pi}$  (2)  $\frac{h}{2\pi}$   
(3)  $\sqrt{2} \frac{h}{2\pi}$  (4) Zero
18. An organic compound has an empirical formula  $\text{CH}_2\text{O}$ , its vapour density is 45. The molecular formula of the compound is.  
(1)  $\text{CH}_2\text{O}$  (2)  $\text{C}_2\text{H}_5\text{O}$   
(3)  $\text{C}_2\text{H}_2\text{O}$  (4)  $\text{C}_3\text{H}_6\text{O}_3$
19. Molecular weight of  $\text{KMnO}_4$  in acidic and neutral medium will be respectively -  
(1) 7 x equivalent wt. and 2 x equivalent wt.  
(2) 5 x equivalent wt. and 3 x equivalent wt.  
(3) 3 x equivalent wt. and 5 x equivalent wt.  
(4) 2 x equivalent wt. and 4 x equivalent wt.
20. At 700 K, the equilibrium constant  $K_p$  for the reaction  $2\text{SO}_3(\text{g}) \rightleftharpoons 2\text{SO}_2(\text{g}) + \text{O}_2(\text{g})$  is  $1.80 \times 10^{-3}$  and  $K_p$  is 14, ( $R = 8.314 \text{ Jk}^{-1} \text{ mol}^{-1}$ ). The numerical value in moles per litre of  $K_c$  for this reaction at the same temperature will be  
(1)  $3.09 \times 10^{-7} \text{ mol-litre}$   
(2)  $5.07 \times 10^{-8} \text{ mol-litre}$   
(3)  $8.18 \times 10^{-9} \text{ mol-litre}$   
(4)  $9.24 \times 10^{-10} \text{ mol-litre}$
21. For the reaction  $\text{CO}(\text{g}) + \text{H}_2\text{O}(\text{g}) \rightleftharpoons \text{CO}_2(\text{g}) + \text{H}_2(\text{g})$  at a given temperature, the equilibrium amount of  $\text{CO}_2(\text{g})$  can be increased by  
(1) Adding a suitable catalyst  
(2) Adding an inert gas  
(3) Decreasing the volume of the container  
(4) Increasing the amount  $\text{CO}(\text{g})$
22. If work done by the system is 300 joule when 100 cal. Heat is supplied to it. The change in internal energy during the process is -  
(1) -200 Joule (2) 400 Joule  
(3) 720 Joule (4) 120 Joule
23. The correct order of increasing C-O bond length of  $\text{CO}$ ,  $\text{CO}_3^{2-}$ ,  $\text{CO}_2$  is  
(1)  $\text{CO}_3^{2-} < \text{CO}_2 < \text{CO}$  (2)  $\text{CO}_2 < \text{CO}_3^{2-} < \text{CO}$   
(3)  $\text{CO} < \text{CO}_3^{2-} < \text{CO}_2$  (4)  $\text{CO} < \text{CO}_2 < \text{CO}_3^{2-}$
24. Bond length of ethane (I), ethene (II), acetylene (III) and benzene (IV) follows the order  
(1) I > II > III > IV (2) I > II > IV > III  
(3) I > IV > II > III (4) III > IV > II > I
25. Molecular shapes of  $\text{SF}_4$ ,  $\text{CF}_4$  and  $\text{XeF}_4$  are  
(1) The same with 2, 0 and 1 lone pairs of electrons respectively  
(2) The same, with 1, 1 and 1 lone pairs of electrons respectively  
(3) Different, with 0, 1 and 2 lone pairs of electrons respectively  
(4) Different, with 1, 0 and 2 lone pairs of electrons respectively
26. The IUPAC name of the compound  is -  
(1) 3,3-Dimethyl cyclohexanol  
(2) 1,1,-Dimethyl-3-hydroxycyclohexane  
(3) 3,3-Dimethyl-1-hydroxycyclohexane  
(4) 1,1-Dimethyl-1-cyclohexanol
27. Nucleophilicity order is correctly represented by  
(1)  $\text{CH}_3^- < \text{NH}_2^- < \text{HO}^- < \text{F}^-$   
(2)  $\text{CH}_3^- = \text{NH}_2^- < \text{HO}^- < \text{F}^-$   
(3)  $\text{CH}_3^- > \text{NH}_2^- > \text{HO}^- > \text{F}^-$   
(4)  $\text{NH}_2^- > \text{F}^- > \text{HO}^- > \text{CH}_3^-$
28. The correct order of reactivity towards the electrophilic substitution of the compounds aniline (I) benzene (II) and nitrobenzene (III) is-  
(1) I < II > III (2) I > II > III  
(3) III > II > I (4) II > III > I
29. Grignard's reagent gives alkane with -  
(1)  $\text{H}_2\text{O}$  (2)  $\text{C}_2\text{H}_5\text{OH}$   
(3)  $\text{C}_2\text{H}_5\text{NH}_2$  (4) All of these
30. Benzene undergoes -  
(1) Nucleophilic substitutions  
(2) Electrophilic substitutions  
(3) Electrophilic additions  
(4) Oxidation reactions in the presence of Lewis acid catalysts

31. Some leaves are removed from the stem cuttings planted for vegetative propagation. This is done
- (1) To increase water uptake
  - (2) Because it helps in rooting of cuttings
  - (3) To reduce water loss
  - (4) Because the cuttings need less food
32. Glomerular filtrate is
- (1) Deproteinised plasma
  - (2) Proteinised plasma
  - (3) Blood plasma
  - (4) Urine stored in urinary bladder
33. Root cap is not found in
- (1) Hollyhock
  - (2) Pistia
  - (3) Sunflower
  - (4) China rose
34. Vasopressin is concerned with
- (1) General metabolism
  - (2) Regulation of heart beat
  - (3) Urine formation
  - (4) Child birth
35. All arteries carry oxygenated blood except
- (1) Systemic
  - (2) Hepatic
  - (3) Pulmonary
  - (4) Cardiac
36. Spleen is
- (1) Haemopoietic
  - (2) Lymphoid
  - (3) Reproductive
  - (4) Celluloid
37. In mammals, the largest vertebra is
- (1) Cervical
  - (2) Lumbar
  - (3) Caudal
  - (4) Sacral
38. Which of the damaged cells cannot be repaired?
- (1) Liver cells
  - (2) Brain cells
  - (3) Bone cells
  - (4) Epidermal cells
39. The major fraction of CO<sub>2</sub> released during cellular respiration is transported by the blood to the lung capillaries
- (1) In combination with haemoglobin
  - (2) As free CO<sub>2</sub>
  - (3) As carbonic acid
  - (4) In the form of bicarbonate ions
40. Legume seeds exhibit dormancy because of
- (1) Undeveloped embryos
  - (2) Hard seed coat
  - (3) Absence of cytokinins
  - (4) Absence of gibberellic acid
41. When a molecule of pyruvic acid is subjected to anaerobic oxidation and forms lactic acid, there is
- (1) Loss of 3 ATP molecules
  - (2) Loss of 6 ATP molecules
  - (3) Gain of 2 ATP molecules
  - (4) Gain of 4 ATP molecules
42. In hydrophytic plants, water and salts are absorbed by
- (1) Roots
  - (2) Leaves
  - (3) Stem
  - (4) Outer layer of plants
43. Which one is not an essential element in plants
- (1) Iron
  - (2) Boron
  - (3) Sulphur
  - (4) Cadmium
44. Pneumatophores occur in plants of
- (1) Sandy soil
  - (2) Saline marshy soil
  - (3) Marshy soil
  - (4) Water
45. Two plants A and B are supplied with CO<sub>2</sub> with H<sub>2</sub>O<sup>18</sup> and CO<sub>2</sub><sup>18</sup> with H<sub>2</sub>O respectively which of the following plant releases O<sup>18</sup> type oxygen in photosynthesis
- (1) A plant
  - (2) B plant
  - (3) Both (1) and (2)
  - (4) First (1) and then (2)

46. If  $a_n = \sum_{r=0}^n \frac{1}{{}^n C_r}$ , the value of  $\sum_{r=0}^n \frac{n-2}{{}^n C_r}$  is :

(1)  $\frac{n}{2} a_n$

(2)  $\frac{1}{4} a_n$

(3)  $na_n$

(4) 0

47. The value of 'c' for which the set,  $\{(x, y) | x^2 + y^2 + 2x \leq 1\} \cap \{(x, y) | x - y + c \geq 0\}$  contains only one point in common is:

(1)  $(-\infty, -1] \cup [3, \infty)$

(2)  $\{-1, 3\}$

(3)  $\{-3\}$

(4)  $\{-1\}$

48. Equation of line passing through mid point of intercepts made by circle  $x^2 + y^2 - 4x - 6y = 0$  on co-ordinate axes is:

(1)  $3x + 2y - 12 = 0$

(2)  $3x + y - 6 = 0$

(3)  $3x + 4y - 12 = 0$

(4)  $3x + 2y - 6 = 0$

49. If z be complex number such that equation  $|z - a^2| + |z - 2a| = 3$  always represents an ellipse then range of  $a (\in R^+)$  is:

(1)  $(1, \sqrt{2})$

(2)  $|1, \sqrt{3}|$

(3)  $(-1, 3)$

(4)  $(0, 3)$

50. The set of possible values of l for which  $x^2 - (\lambda^2 - 5\lambda + 5)x + (2\lambda^2 - 3\lambda - 4) = 0$  has roots, whose sum and product are both less than 1, is

(1)  $\left(-1, \frac{5}{2}\right)$  (2)  $(1, 4)$

(3)  $\left[1, \frac{5}{2}\right]$  (4)  $\left(1, \frac{5}{2}\right)$

51. If  $\alpha, \beta$  be the roots of  $4x^2 - 16x + \lambda = 0$ , where  $\lambda \in R$ , such that  $1 < \alpha < 2$  and  $2 < \beta < 3$ , then the number of integral solutions of  $\lambda$  is

(1) 5

(2) 6

(3) 2

(4) 3

52. Let  $\alpha, \beta, \lambda$  be the roots of  $(x - a)(x - b)(x - c) = d, d \neq 0$ , then the roots of the equation

$(x - \alpha)(x - \beta)(x - \gamma) + d = 0$  are :

(1)  $a + 1, b + 1, c + 1$

(2)  $a, b, c$

(3)  $a - 1, b - 1, c - 1$

(4)  $\frac{a}{b}, \frac{b}{c}, \frac{c}{a}$

53. There are n straight lines in a plane, no two of which are parallel, and no three pass through the same point. Their points of intersection are joined. Then the number of fresh lines thus obtained is:

(1)  $\frac{n(n-1)(n-2)}{8}$

(2)  $\frac{n(n-1)(n-2)(n-3)}{6}$

(3)  $\frac{n(n-1)(n-2)(n-3)}{8}$

(4) None of these

54. If the straight lines joining the origin and the points of intersection of the curve  $5x^2 + 12xy - 6y^2 + 4x - 2y + 3 = 0$  and  $x + ky - 1 = 0$  are equally inclined to the x-axis, then the value of k is equal to :

(1) 1

(2) -1

(3) 2

(4) 3

55. The general solution of the equation  $\cos x \cdot \cos 6x = -1$ , is :
- (1)  $x = (2n+1)\pi$
  - (2)  $x = 2n\pi, n \in \mathbb{I}$
  - (3)  $x = (2n-1)\pi, n \in \mathbb{I}$
  - (4) none of these
56. The equation  $\cos \theta = x + \frac{p}{x}$  for all  $x \in \mathbb{R}$  has a real solution for  $\theta$ . Then :
- (1)  $p = \frac{1}{2}$
  - (2)  $p \leq \frac{1}{4}$
  - (3)  $p \geq \frac{1}{4}$
  - (4) None of these
57. If  $x + y = z$ , then  $\cos^2 x + \cos^2 y + \cos^2 z - 2 \cos x \cos y \cos z$  is equal to
- (1)  $\cos^2 z$
  - (2)  $\sin^2 z$
  - (3)  $\cos(x + y - z)$
  - (4)  $\cos^2 x$
58. If  $a\mathbb{N} = \{ax : x \in \mathbb{N}\}$  and  $b\mathbb{N} \cap c\mathbb{N} = d\mathbb{N}$ , where  $b, c \in \mathbb{N}$  are relatively prime, then
- (1)  $d = bc$
  - (2)  $c = bd$
  - (3)  $b = cd$
  - (4) None of these
59. If  $f(y) = \log y$ , then  $f(y) + f\left(\frac{1}{y}\right)$  is equal to
- (1) 2
  - (2) 1
  - (3) 0
  - (4) -1
60. If  $f : \mathbb{R} \rightarrow \mathbb{R}$  satisfies  $f(x+y) = f(x) + f(y)$ , for all  $x, y \in \mathbb{R}$  and  $f(1) = 7$ , then  $\sum_{r=1}^n f(r)$  is
- (1)  $\frac{7n}{2}$
  - (2)  $\frac{7(n+1)}{2}$
  - (3)  $7n(n+1)$
  - (4)  $\frac{7n(n+1)}{2}$

## ANSWERS (RADIANT - CLASS 12TH)

### PHYSICS

Q	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15
Ans	1	4	4	1	1	1	2	2	3	3	2	2	3	3	2

### CHEMISTRY

Q	16	17	18	19	20	21	22	23	24	25	26	27	28	29	30
Ans	1	4	4	2	1	4	1	4	3	4	1	3	2	4	2

### BIOLOGY

Q	31	32	33	34	35	36	37	38	39	40	41	42	43	44	45
Ans	3	1	2	3	3	2	2	2	4	2	1	3	4	2	1

### MATHS

Q	46	47	48	49	50	51	52	53	54	55	56	57	58	59	60
Ans	4	4	4	4	4	4	2	3	2	3	2	7	1	3	4